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Stoichiometry Example $C_6H_6 + Br_2 \rightarrow C_6H_5Br + HBr$ Benzene (C_6H_6) reacts with Bromine to produce bromobenzene (C_6H_5Br) and hydrobromic acid. If 30. g of benzene reacts with 65 g of bromine and produces 56.7 g of bromobenzene, what is the percent yield of the reaction? 30.g 65 g 56.7 g ...

Chapter 3 Stoichiometry - Chemistry

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