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## OBJECTIVES

### **PPT - Chapter 8 (Modern Chemistry) PowerPoint presentation ...**

CHAPTER 8 REVIEW Chemical Equations and Reactions MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. b A balanced chemical equation represents all the following except (a) experimentally established facts. (b) the mechanism by which reactants combine to form products.

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SECTION 8-3 SHORT ANSWER Answer

the following questions in the space

provided. 1. List four metals that will not replace hydrogen in an acid.

## **Modern Chemistry Chapter 8 Section 3 Review Answers**

CHAPTER 8 REVIEW Chemical Equations and Reactions SECTION 3 SHORT

ANSWER Answer the following questions in the space provided. 1. List four metals that will not replace hydrogen in an acid.

Choose from Cu, Ag, Au, Pt, Sb, Bi, and Hg.

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CHAPTER 8 REVIEW Chemical Equations and Reactions SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. List four metals that will not replace hydrogen in an acid. Choose from Cu, Ag, Au, Pt, Sb, Bi, and Hg. 2. Consider the metals iron and silver, both listed in Table 3on page 286 of the text. Which one

## **8 Chemical Equations and Reactions**

4. a. Use five to six data points from

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## Answers

Appendix Table A-8 in the text to sketch the curve for water vapor's partial pressure versus temperature on the graph provided below.

Temperature (°C)	Partial Pressure (torr)
10	10
15	15
20	20
25	25
30	30

Do the data points lie on a straight line?

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Measurements and Calculations  
SECTION 1 SHORT ANSWER Answer the  
following questions in the space  
provided. 1. Determine whether each of  
the following is an example of  
observation and data, a theory, a  
hypothesis, a control, or a model.

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1.7 Chapter Homework: 1.8 References:  
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