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Types of Reactions. chemistry practical class 10 Types of Reactions Introduction

> Chemical reaction: The reaction in which two or more substances react to form a new product with new properties is called a chemical reaction. > Chemical reaction shows following properties:

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Experiment 2. > Chemical formula of hydrochloric acid is HCl. > When it is dissolved in water; releases H^+ (aq) ions, these H^+ ions cannot exist alone. Hence, it combines with water to form H_3O^+ (aq) ions.

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This CBSE Class 10 Science lab manual

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familiarises students with the available instruments in the laboratory and informs them about the do's and don't while performing experiments. Teachers also explain to students about all these things at the beginning of the session.

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of concepts. All the observations are noted down while performing the Class 10 Science Lab Manual experiments.

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NCERT Class 10 Science Lab Manual - Reactivity Series. > The atoms of the elements in the periodic table react with each other to form new compounds (except noble gases). The atoms react due to its tendency to lose, gain or share electrons.

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laboratory is important for making the study complete, especially for a subject like Science and Chemistry. CBSE has included the practicals in secondary class intending to make students familiarised with the basic tools and techniques used in the labs. With the help of this, they can successfully perform the experiments listed in the CBSE Chemistry Lab Manual.

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CBSE Class 10 Science Lab Manual - Soap Preparation. Aim. To study saponification reaction for the preparation of soap. Materials Required. 20% sodium hydroxide solution, castor

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oil, common salt, filter paper, distilled water, litmus paper (red and blue) strips, glass rod, borosil beakers, Bunsen burner, wire gauze and tripod stand.

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NCERT Class 10 Science Lab Manual - Properties of Acetic Acid. Introduction. > Acetic acid is a member of carboxylic acid group and it's IUPAC name is ethanoic acid. > Acidic property of CH_3COOH is due to its ability to produce hydrogen ions (H^+) in water. > The members of the homologous series of the carboxylic acids are found in fruits and in vinegar.

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