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HL Paper 2. Enthalpy changes depend on the number and type of bonds broken and formed. ... Two chemistry students wished to determine the enthalpy of hydration of anhydrous magnesium sulfate. They measured the initial and the highest temperature reached when anhydrous magnesium sulfate, $\text{MgSO}_4(\text{s})$, was ...

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«signal at» 2.2 2.7 «ppm» AND «H atom/proton of» RCOCH₂- present OR no «signal at» 4.5 6.0 «ppm» AND absence of «H atom/proton next to» double bond/C=C Accept a specific value or range of wavenumbers and chemical shifts. Accept "two signals with areas 1:3". 2 (continued...)

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Subject Details: Chemistry HL Paper 2 Markscheme Mark Allocation Candidates are required to answer ALL questions in Section A [40 marks] and TWO questions in

MARKSCHEME

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IB Chemistry HL 31 Common Mistakes: Richard Thornley, the author of this video, has several other helpful videos on IB Chemistry SL and HL available for free on YouTube IB Chemistry Web: This site goes over the syllabus in-depth and explains key definitions and facts you need to know.

The Best IB Chemistry Study Guide and Notes for SL/HL

Pretty much the entire SL and HL curriculums are covered in the various playlists he has. The resource is completely free is it's just a YouTube channel. The downside of this is that the resource is solely videos (no IB Exam Questions or Worksheets), so Past Paper practice is still required. Pros: Huge range of free Chemistry Videos. High ...

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HL Paper 2. An acidic sample of a waste solution containing Sn^{2+} ... Calculate the Gibbs free energy, ... An acidified solution of potassium dichromate is often used as an oxidizing agent in organic chemistry.

HL Paper 2 - GitHub Pages

There are three papers: Paper 1 (multiple choice) - 20%; Paper 2 (structured questions) - 40%; Paper 3 (data and options) - 20%; Total - 80%; Paper 1 (SL): 45 minutes. 30 multiple-choice questions on core, about 15 of which are common with HL. The questions on paper 1 test assessment objectives 1, 2 and 3. The use of calculators is not permitted.

IB Chemistry- Assessment in the Diploma Program

$2.4 \times 10^{-9} > 1.4 \times 10^{-16}$ AND precipitate will form If value given is used: ALTERNATIVE 3: $\llbracket \text{Pb}^{2+} \rrbracket \llbracket \text{OH}^- \rrbracket^2 2.4 \times 10^{-4} (1.0 \times 10^{-2})^2 \gg 2.4 \times 10^{-8} Q > K_{sp}$ AND precipitate will form OR $2.4 \times 10^{-8} > 1.43 \times 10^{-20}$ AND precipitate will form 2

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