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thermal reservoir at constant temperature T h, converts an amount W of this heat into work, and expels the remainder Q c into another thermal reservoir at a constant temperature T c. The ef?ciency e of the heat engine is the fraction of Q h that is converted into work, so e= W Q h = Q h Q c Q h =1 Q c Q h (1) Since the absorbed heat Q

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This agrees with Schroeder's Figure 1.14. To put in some numbers, we can use the data from the earlier problem. For water, the values given by Schroeder are (at 25 C =298 K): = 2:57 10 4 K 1 (20) T = 4:52 10 10 Pa 1 (21) For one mole of water, the volume is 18:068 10 6 m3, so C P C V = (298) 18:068 10 6 2:57 10 4 2 4:52 10 10 (22) = 0:787 J K 1 (23) C

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