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Chapter 22: Nuclear Chemistry Section 22-1: The Nucleus • Atomic nuclei= protons and neutrons (together are nucleons) o Nuclide= an atom—identified by # of protons/neutrons/electrons o Caused by conversion of mass to energy when nucleus forms Nuclear ...

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Chapter 23 Nuclear Chemistry Notes 1 CHAPTER 23 NUCLEAR CHEMISTRY 23.1 THE NATURE OF NUCLEAR REACTIONS radioactivity - the spontaneous decay of an unstable nucleus with accompanying emission of radiation. nuclide - atom with a specific number of protons and neutrons in its nucleus.

There are 271 stable nuclides in nature, others are radioactive are radioactive.

CHAPTER 23 NUCLEAR CHEMISTRY

KEY: Radioactive isotope MSC: 2 13 Nuclear chemistry sec 1 answer key. ANS: C A radioisotope with an atomic number above 83 attains stability by reducing the element within the band of stability. Chapter 6 Practice Test. Nature of Science. Nuclear Answer Section.

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What is the energy released in this beta+ nuclear reaction: 24 13 Al arrow 24 12 Mg + 0 1 e? (The atomic mass of 24Al is 23.999939 u and that of 24Mg is 23.985041 u) View Answer

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